

Global Bromine Industry And Its Outlook

- Production by Country, Production Process, Application and Forecast.

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1. Bromine Production by Country 2. Concentration by Source 3. Bromine Production Capacity by Region **4. Bromine Production Flow** 5. Bromine Compounds Market by Major Application 6. Bromine Flame Retardants

7. Bromine Outlook

Bromine: the essentials

Brief description: bromine is the only liquid nonmetallic element. It is a member of the halogen group. It is a heavy, volatile, mobile, dangerous reddish-brown liquid. The red vapor has a strong unpleasant odor and the vapor irritates the eyes and throat. It is a bleaching. When spilled on the skin it produces painful sores. It is a serious health hazard, and maximum safety precautions should be taken when handling it. Table: basic information about and classifications of bromine. Name: Bromine Symbol: Br Atomic number: 35 Atomic weight: 79.904 (1) Standard state: liquid at 298K **CAS Registry ID: 7726-95-6**



Isolation

Isolation

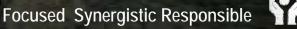
Isolation: bromine is available commercially so it is not normally necessary to make it in the laboratory. Bromine also occurs in seawater as the sodium salt but in much smaller quantities than chloride. It is recovered commercially through the treatment of seawater with chlorine gas and flushing through with air. In this treatment, bromide is oxidized to bromine by the chlorine gas. The principle of oxidation of bromide to bromine is shown by the addition of a little chlorine water to aqueous solutions of bromide. These become brown as elemental bromine forms.

$2Br + Cl2 \rightarrow 2Cl + Br2$

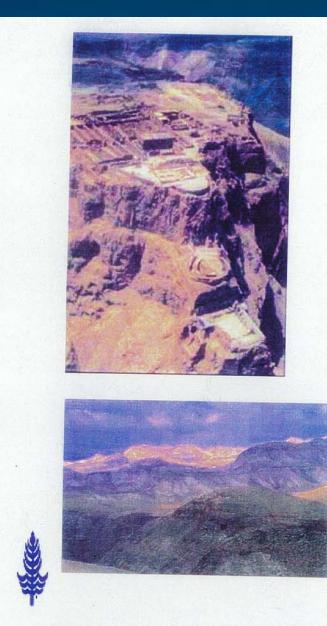
Small amounts of bromine can also be made through the reaction of solid sodium bromide, NaBr, with concentrated sulphuric acid, H2SO4. The first stage is formation of HBr, which is a gas, but under the reaction conditions some of the HBr is oxidized by further H2SO4 to form bromine and sulphur dioxide. This reaction does not work with the corresponding chlorides and fluorides.

NaBr (s) + H2SO4 (I) \rightarrow HBr (g) + NaHSO4 (s)

2HBr (g) + H2SO4 (I) → Br2 (g) + SO2 (g) + 2H2O (I)



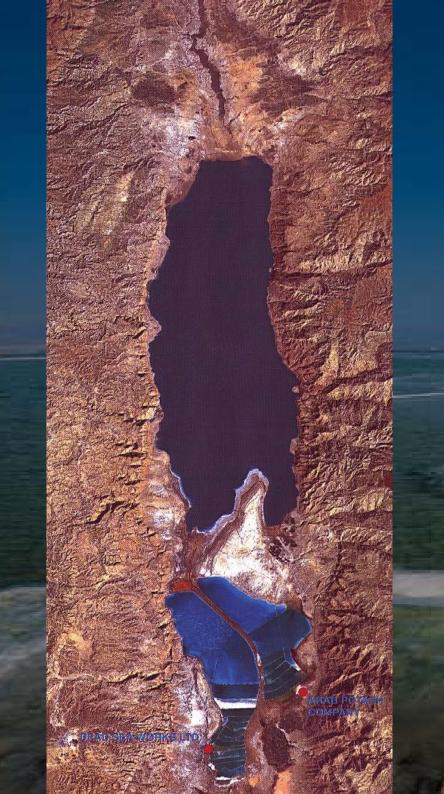




Israel: views from the Negev region



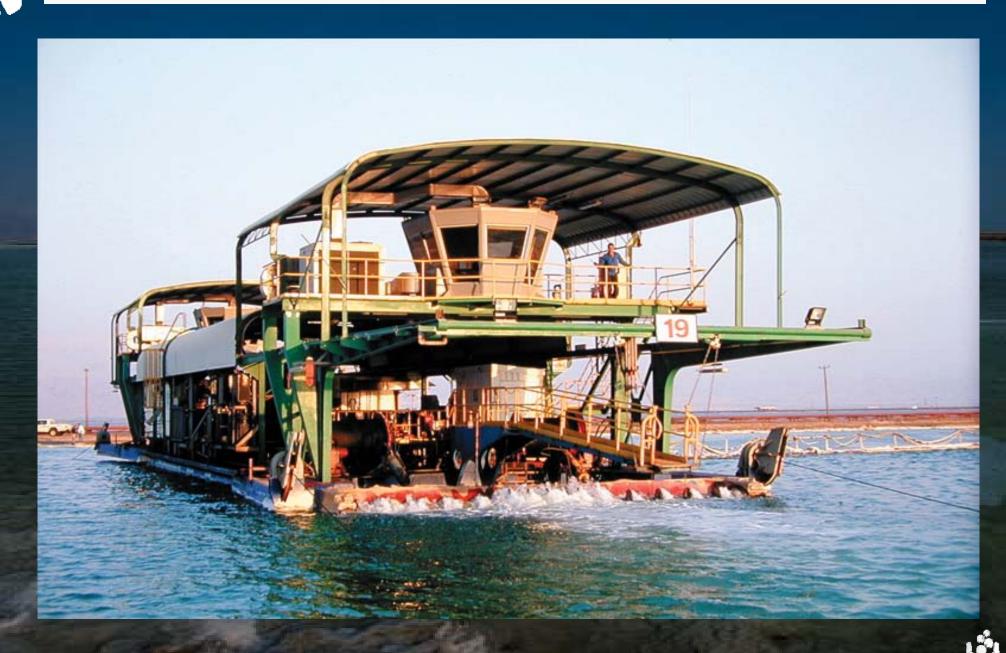
May 1999



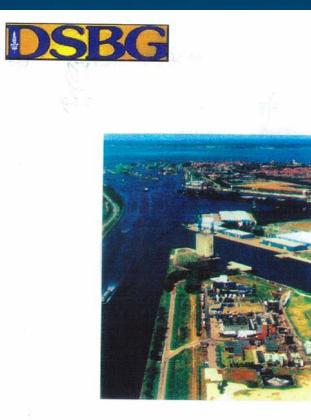
The Dead Sea



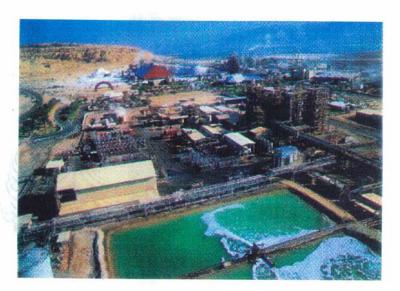
View of Dead Sea area Landsat TM3,2,1 (R.G.B) docugraph INFORMATION SYSTEMS Raw material harvester







Bromine Compounds Plant - Netherlands



Bromine/Chlorine Plant - Israel

An Introductory Presentation

23/11/00

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BROMME ISOTANKS ROUNDTRIP

MPROVEMENT





Where Has the Market Share Gone? To China...



Aerial photo of Bromine production area in Shandong

Chinese Bromine production and packing facility

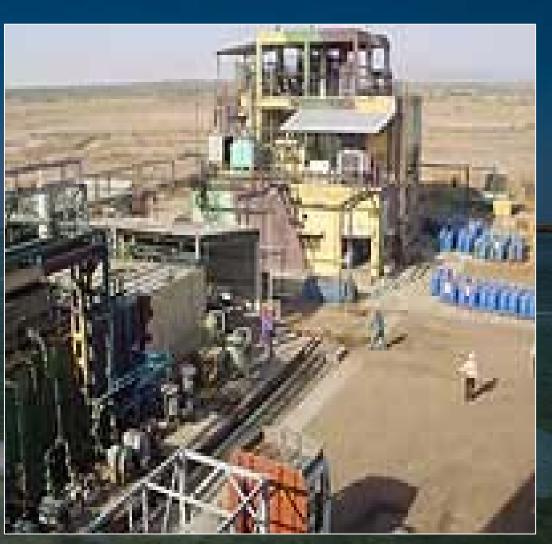
Focused Synergistic Responsible





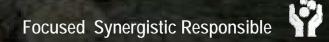
Bromine Production - India





Indian evaporation pond

Indian plant



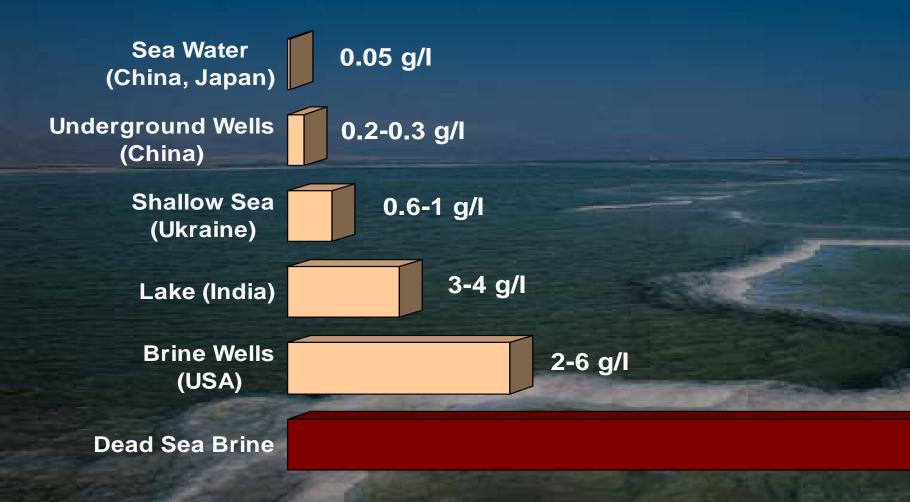




Satellite photo of relevant Ukrainian regions

Ukrainian bromine plant





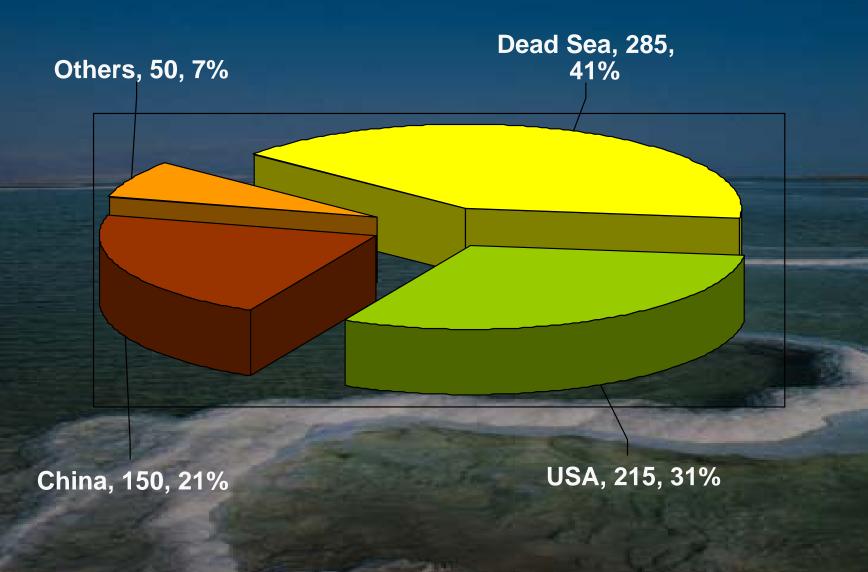
Source: ICL estimates

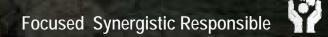
Focused Synergistic Responsible

10-12 g/l



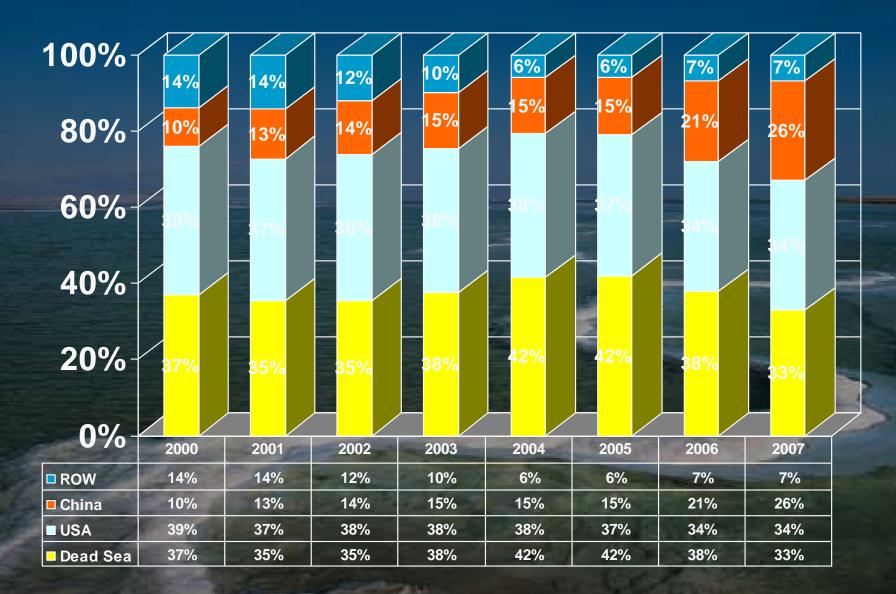
Global Bromine Production Capacity- by Region (700 KMT)





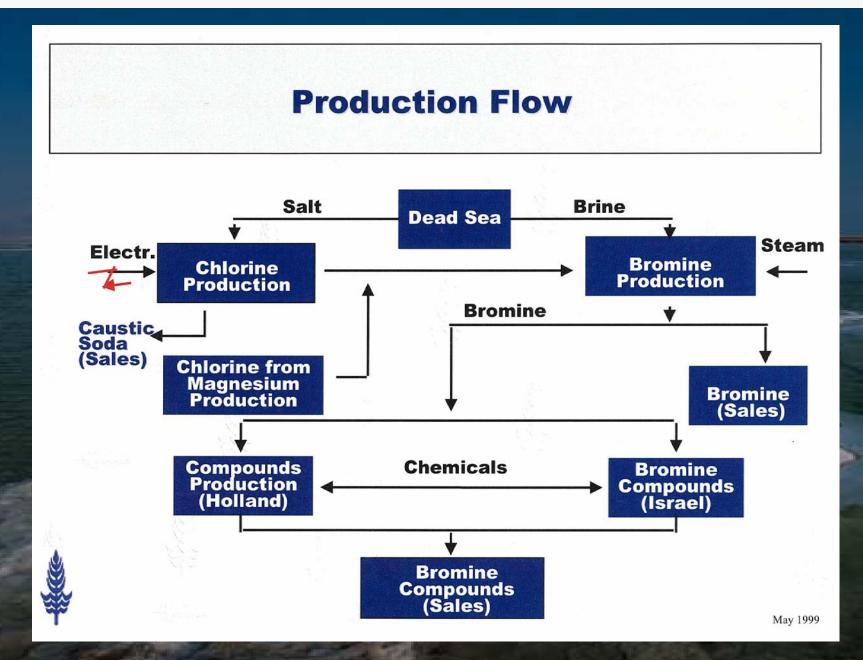


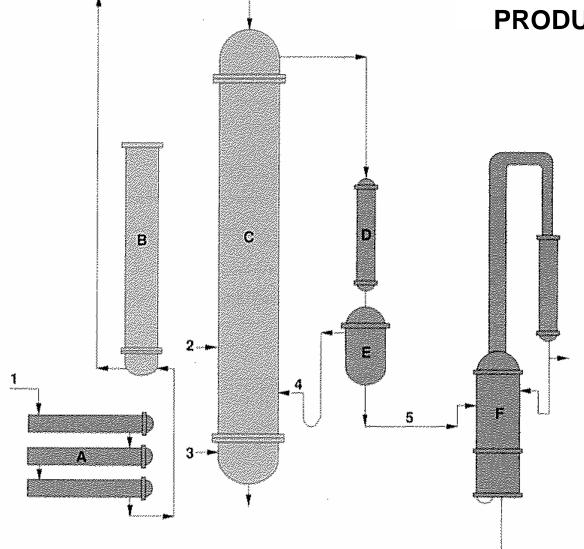
Bromine Production (by Region - % of total production)





Production Flow in ICL-IP





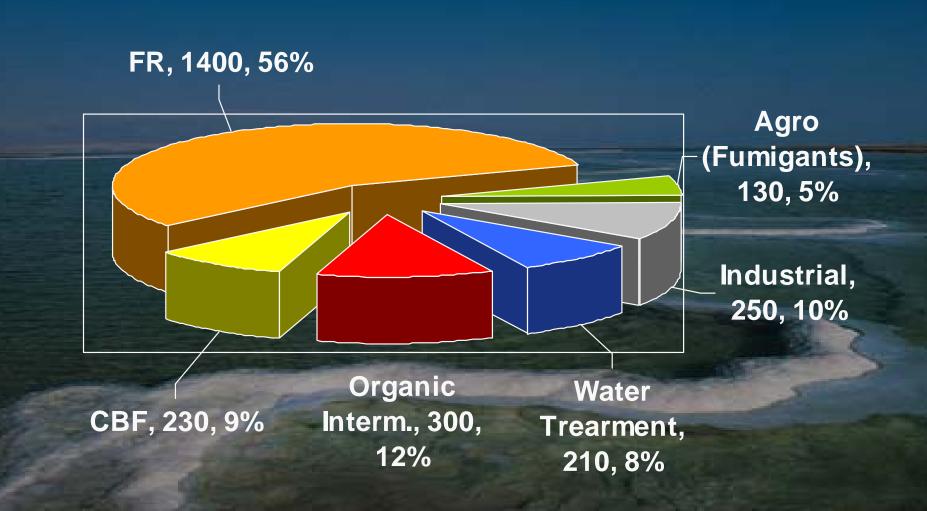
PRODUCTION FLOW CHART

LEGEND: Bromideenriched Dead Sea brine (1), heated by preheaters (A) and heat-exchanger (B), flows into a reaction column (C) in which the introduced chlorine (2) liberated elemental bromine. Steam (3) is used to strip the bromine out of the reaction column and into a condenser (D). from which the bromine flows into a phase separator (E). The bromine-saturated water (4) is returned to the reaction column, while the bromine (5) passes, first, to a distillation column (F). After cooling, the purified liquid bromine flows to storage (6),

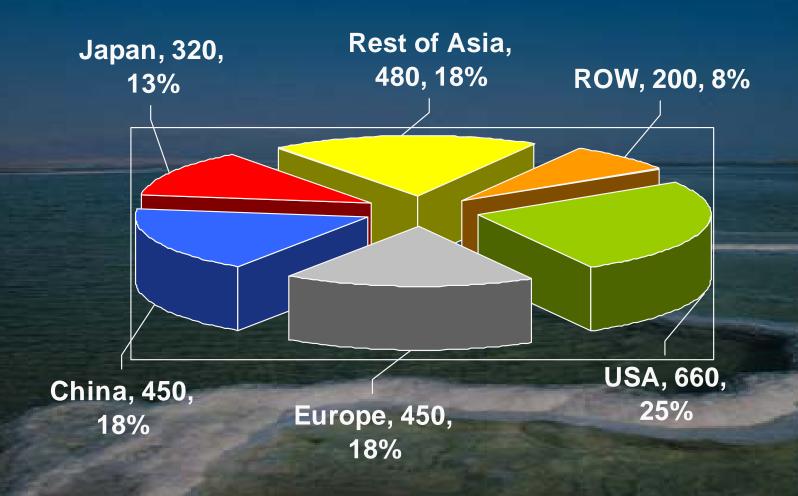
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Global Bromine Compounds Market- by Major Application (Total in 2007- \$2.5 BN)



Bromine Compounds Consumption – By Region (\$2.5 BN - 2007)



Demand by Application (In terms of Bromine – 575 KMT in 2007)

Bromine

Agro-Fumigants	25	4%
Industrial	105	18%
Organic Interm	63	11%
CBF		
Water Treatment	32	5%
FR	283	49%

Use of Bromine Compounds

- Flame retardants
- Agrochemicals
- Water treatment chemicals
- Pharmaceuticals and cosmetics
 - Lead scavenger in anti-knock fuels
 - Oilfield completion fluids
 - Photographic chemicals
 - Desiccants for cooling systems
 - Bleaching and oxidizing chemicals
 - Dyestuffs
 - Monomers for specialty polymers
 - **Precision cleaning**



Electrical and electronic continues to improve the quality of our lives























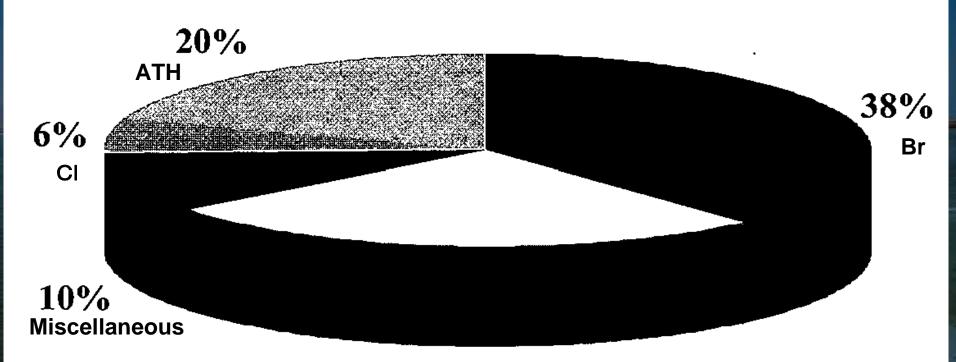


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Flame Retardants Worldwide Market

2,700 MM\$ in 2005 – Bromine 38%



26% р

Miscellaneous include : inorganic phosphates, melamine, MG(OH)2, boron and Molybdenum compounds



February 14th 2005

100m high Windsor Tower in Madrid catching fire and

completely destroyed.







Focused Synergistic Responsible



The Background to VECAP Why use Flame Retardants?

Every day in Europe :

12 fire victims and 120 severely injured 75 % of victims are in their homes

Economic damage : +/- 25 billion Euro's per year

In the UK since 1988 furniture regulation: 3200 lives saved 1988-2000 29000 less non-fatal injuries



Flame retardants and other technologies are a critical part of the fire safety solution. The fire on an old (1980's) European TV self-extinguishes, while the fire in a non fire-safe TV, recently sold in Europe, keeps burning and growing



2, 3, 7, 8, - TBDD 1, 2, 3, 7, 8, - PeBDD 2, 3, 7, 8, - TBDF 2, 3, 4, 7, 8, - PeBDF 4 together : 1 ppb max. Above 4 plus below 4 = Total 8 : 5 ppb max. 1, 2, 3, 4, 7, 8, - HxBDD 1, 2, 3, 7, 8, 9, - HxBDD 1, 2, 3, 6, 7, 8, - HxBDD 1, 2, 3, 7, 8, - PeBDD



Br-FR by RESIN APPLICATION IN JAPAN

Br-FR

ABS

RESIN

Deca,TBBA,Br-PC Olygomer, TAIC-6B,BEO,Br-PS, Ethylene-Bis(Tetrabromphtalimide) Decabromodiphenylethane, Tris(TBNPA)phosphate Br-Aromatic Triazine

EPS

TBBA-Bis(Allylether),HBCD TBP-Allylether,Tetrabromethane HIPS Deca,TBBA,HBCD, Ethylene-Bis(Tetrabromphtalimede) BEO,Tris(TBNPA)phosphate Decabromodiphenylethane Br-Aromatic Triazine TBBA-Bis(2,3Dibrompropylether)





PP

TBBA-Bis(2,3Dibrompropylether) Deca,TBBS,HBCD TBBS-Bis(2,3Dibrompropylether) Decabromdiphenylethane Tris(TBNPA)phosphate Br-Aromatic Triazine

PE

Deca, Decabromdiphenylethane TBBS,TAIC-6B TBBS-Bis(2,3Dibrompropylether)

PC TBBA, Br-PC Olygomer, BEO

PMMA Tris(TBNPA)phosphate PBB-MA,PBB-PA TAIC-6B, BEO TPB-EO-Acrylate





PBT

Deca,TBBA-2EO,TBBA-4EO Br-PC Olygomer,PBB-PA BEO Ethylene-bis(Tetrabromphtalimide) Decabromodiphenylethane

PET Deca,Br-PC Olygomer,BEO Br-PS,TAIC-6B Ethylene-Bis(Tetrabromphtalimide) Decabromdiphenylethane

NYLON Deca,Poly-Dibromphenyleneoxide Br-PS,BEO,Br-PC Decabromdiphenylethane Ethlene-Bis(Tetrabromphtalimide)

PVCDecaEPDMDeca,Decabromdiphenylethane



EPOXY (laminate, BEO,)

Deca,HBB,TBBA,TBP,BEO, Br Aromatic Triazine Dibromcrezylglycidylether Dibromphenylglycidylether

Phenol

TBBA, TBP, BEO

DBNPG,TBNPA,TBBA-2EO

URETHANE ELASTOMER Deca,BEO

UPET

TBBA-2EO, DBNPG, TBNPA BEO

DCPD Silicone (Rubber) Deca, Br-PS

Deca, Decabromdiphenlyethane

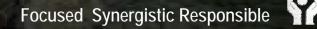


TEXTILE (Acrylic,SBR,PP,Polyester,Cotton)

Deca,HBCD,Pentabromtoluene TBBA-2EO Tris(TBNPA)phosphate

Adhesive,Adhesive Tape,Paper,Paint,Ink Plywood

> Deca,TBBA-2EO,HBCD Ammoniumbromide Tris(TBNPA)phosphate Decabromdiphenylethane



Outlook

The demand for bromine based products will grow at an average rate of 2-2.5% per year (2% in the first 3 years and later 2.7%)

Major drivers and growth areas:

Adoption of fire safety regulation.

New applications (Mercury reduction,)

Organic growth in the various applications, following growth of the global economy.

Offset by:

Continued phased-out of MBr

Decline of Deca and HBCD

Emergence of non-halogen substitutes for BFR's.

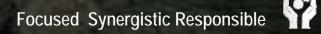


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